



What are gel materials used for? These gel materials have successfully served as electrode materials, electrolytes, self-supported current collectors, 3D binder systems, etc. in various kinds of energy conversion and storage applications, such as lithium ion batteries, supercapacitors, catalysts, and fuel cells.



Can gel materials be used for energy applications? In the past decades,great progress has been achieved in the development of gel materials for energy applications,and several review papers have been published that have focused on specific materials,such as carbon-based gels ,conductive polymer gels ,and gel electrolytes .



What conductive gel materials are available for energy applications? In this review, we provide a full picture of the state-of-the-art gel materials that are available for energy applications and discuss various electrically conductive gel materials, including carbon-based gels, conductive polymer gels, as well as ionically conductive gels.



Are gel-based nanomaterials a promising material platform for Advanced Energy Applications? Recent development of gel-based nanomaterials including carbon based gels, conductive polymer gels, ionic gels and inorganic gels is reviewed. Electronically/ionically conductive gels build up a promising material platform for advanced energy applications.



Can 3D gel materials be used for energy conversion & storage? Although gel materials with 3D network structures have been synthesized using various inorganic materials and employed in applications such as catalysis,oil removal,and dye absorption,few studies on their application for energy conversion and storage have been reported.



Are gel electrolytes suitable for flexible energy storage systems? Recently reported gel electrolytes for flexible energy storage systems with their application and properties. Disclaimer/Publisher???s Note: The statements, opinions and data contained in all publications are solely



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Gel materials find diverse applications as electrode materials, electrolytes, self-supported current collectors, skeletons of active materials, and 3D binders in the field of energy conversion and storage, such as electrochemical CO 2 reduction to value-added products, oxygen reduction reactions, and the oxygen evolution reaction in metal???air



Materials possessing these features offer considerable promise for energy storage applications: (i) 2D materials that contain transition metals (such as layered transition metal oxides 12



select article Corrigendum to "Multifunctional Ni-doped CoSe₂ nanoparticles decorated bilayer carbon structures for polysulfide conversion and dendrite-free lithium toward high-performance Li-S full cell" [Energy Storage Materials Volume 62 (2023) 102925]



Hydrogel is an ideal material for flexible electrochemical energy storage components due to its good conductivity and softer texture, which is expected to promote electrochemical energy storage technology toward high efficiency, durability, environmental protection, etc., and expand the application range.



Now in many types of gels, as a kind of new advanced materials, the ILs-based gels which means that the gel contains ILs are attractive. ILs are organic salts formed by organic cations together with organic or inorganic anions with melting points below 100 ?C and have been applied to prepare some gels [[16], [17], [18]].Poly(ionic liquids) (PILs) are polymer chains ???





Hydrogels are soft materials that consist of physically or chemically cross-linked polymer networks and a large quantity of water. Hydrogels have a high water content and low elastic modulus (~100



The kinetic analysis of the dehydration reaction of silica gel showed that the average activation energy for the desorption of silica gel is 66.75 kJ/mol. Compared with the activation energy for the reaction of the energy-storage materials listed in Table 4, the activation energy for the desorption of silica gel is small and the dehydration



lonic liquids (ILs) are molten salts that are entirely composed of ions and have melting temperatures below 100 ?C. When immobilized in polymeric matrices by sol???gel or chemical polymerization, they generate gels known as ion gels, ionogels, ionic gels, and so on, which may be used for a variety of electrochemical applications. One of the most significant ???



Electrolytes have played critical roles in electrochemical energy storage. In Li-ion battery, liquid electrolytes have shown their excellent performances over decades, such as high ionic conductivity (?? 1/4 10???3 S cm???1) and good contacts with electrodes. However, the use of liquid electrolytes often brought risks associated with leakage and combustion of organic ???



Electrochemical energy storage devices, such as lithium ion batteries (LIBs), supercapacitors and fuel cells, have been vigorously developed and widely researched in past decades. However, their safety issues have appealed immense attention. Gel electrolytes (GEs), with a special state in-between liquid and solid electrolytes, are considered as the most ???





The 2D allotrope of carbon-based material is an ideal candidate for next generation energy devices. This chapter gives an overview on the recent research on graphene-incorporated sol-gel materials for energy conversion and storage applications, such as supercapacitors, solar cells, lithium-ion batteries, and fuel cells.



Electronic conductive gels hold great promise for energy conversion and storage applications, such as batteries, supercapacitors, and fuel cells, owing to their robust mechanical strength, ???



Gelatin is a biological macromolecular material with sol-gel properties containing lots of active functional groups [68]. The mechanical properties of gelatin hydrogels are relatively poor. Therefore, there will be necessary to combine with other degradable materials to ensure that the energy storage and conversion system can match the



In the current era, national and international energy strategies are increasingly focused on promoting the adoption of clean and sustainable energy sources. In this perspective, thermal energy storage (TES) is essential in developing sustainable energy systems. Researchers examined thermochemical heat storage because of its benefits over sensible and latent heat ???



Polymers obtained from biomass are promising alternatives to petro-based polymers owing to their low cost, biocompatibility, and biodegradability. Lignin, a complex aromatic polymer containing several functional hydrophilic and active groups including hydroxyls, carbonyls, and methoxyls, is the second most abundant biopolymer in plants. In particular, sustainable lignin ???





The energy storage mechanism of secondary batteries is mainly divided into de-embedding (relying on the de-embedding of alkali metal ions in the crystal structure of electrode materials to produce energy transfer), and product reversibility (Fig. 5) (relying on the composite of active material and conductive matrix, with generating and



School of Materials Science and Engineering, University of New South Wales, Sydney 2052, Australia This Special Issue on "Gel Polymer Electrolytes for Energy Storage" is dedicated to recent developments from theoretical and fundamental aspects to the synthesis, characterization, and applications of gel polymer electrolytes.



To determine the prepared materials" thermal energy storage performance, 2???6 g of each sample was tested in a lab-scale apparatus. F. H. Water vapor adsorption in silica gel for thermal



Compared with traditional liquid electrolytes, gel polymer electrolytes (GPEs) are preferred due to their higher safety and adaptability to the design of flexible energy storage devices. This review summarizes the recent progress of GPEs with enhanced physicochemical properties and specified functionalities for the application in



The hybrid sol-gel materials had shown potential for efficient dielectric energy storage because of their high orientational polarization under an electric field, so the group decided to pursue



There is widespread recognition that the use of energy in the twenty-first century must be sustainable. Because of its extraordinary flexibility, silica sol???gel chemistry offers the opportunity to create the novel materials and architectures which can lead to significant advances in renewable



energy and energy storage technologies. In this paper, we review some of the ???





Using a hybrid silica sol-gel material and self-assembled monolayers of a common fatty acid, researchers have developed a new capacitor dielectric material that provides an electrical energy



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In this study, a series of gelators (Gn, n is the number of carbon atoms of used fatty alcohol, n = 2, 4, 6, 8, 10, 12, 14, 16 and 18) were synthesized by reacting 4,4???-diphenylmethane diisocyanate with fatty alcohols. Meanwhile, n-octadecane-based gels as form-stable phase change materials (FSPCMs) for thermal energy storage were prepared by ???



Among various energy storage technologies, energy storage based on phase change materials (PCMs) is conducted through the absorption, storage and release of heat in the phase transition process. PCM as the key working medium is a material with non-corrosive, energy-saving and stable physical properties [1], which also presents the advantages



Gels are attracting materials for energy storage technologies. The strategic development of hydrogels with enhanced physicochemical properties, such as superior mechanical strength, flexibility



Gels are attracting materials for energy storage technologies. The strategic development of hydrogels with enhanced physicochemical properties, such as superior mechanical strength, flexibility, and charge transport capabilities, introduces novel prospects for advancing



next-generation batteries, fuel cells, and supercapacitors.





The types of energy storage materials are mainly divided into sensible heat storage materials, latent heat storage materials and chemical heat storage materials. The products are easy to crack. Sol-gel derived materials are applied in optics, electronics, energy, ceramics, forging, sensors, medicine and separation chromatography [154].